

# Student Exploration Solubility And Temperature Answers

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## **Student Exploration Solubility And Temperature**

In the Solubility and Temperature Gizmo™, you will study how temperature affects how much solute will dissolve in a solution. To begin, check that Potassium nitrate is selected and the Temp. of the water is 20 °C. Click OK.

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Solubility and Temperature Gizmo : ExploreLearning Add varying amounts of a chemical to a beaker of water to create a solution, observe that the chemical dissolves in the water at first, and then measure the concentration of the solution at the saturation point.

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The Solubility and Temperature Gizmo allows students to measure the effect of temperature on solubility of two different substances, including salt, by adding varying amounts of a chemical to a beaker of water to create a solution and then measuring the concentration of

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## **William Edwin Rudge**

Student Exploration: Solubility and Temperature. Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent. Prior Knowledge Questions (Do these BEFORE using the Gizmo.) What happens when you stir a spoonful of sugar into hot water?

## **Student Exploration Sheet: Growing Plants**

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## **Solubility and Temperature - Mrs. Ashley's Science Spot**

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## **Solubility And Temperature Gizmo Answer Key Activity A**

Gather data: Use the Gizmo to measure the solubility of potassium nitrate at each temperature given in the table below. Then, graph the resulting solubility curve at right. Temperature Solubility (g/100 mL)

10 °C	20.54
20 °C	30.65
30 °C	44.44
40 °C	61.91
50 °C	83.06
60 °C	107.89
70 °C	136.4
80 °C	

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168.59 90 °C 204.47 3.

## **A How much sodium chloride did you add 40 B ... - Course Hero**

This lab-based lesson is designed to help students experience the fact that how and whether a solution can be made is determined in large part by the chemical composition of the solute and solvent. How fast the solution can be made is determined in part by the temperature at which the mixture is created. Studying solubility also gives students another lens with which to peer into the nanoscale.

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