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E1-26 The room volume is $(21 \times 13 \times 12)\text{ft}^3(0.3048 \text{ m/ft})^3 = 92.8\text{m}^3$. The mass contained in the room is $(92.8\text{m}^3)(1.21\text{kg/m}^3) = 112\text{kg}$: E1-27 One mole of sugar cubes would have a volume of $N_A \times 1.0 \text{ cm}^3$, where N_A is the Avogadro constant. Since the volume of a cube is equal to the length cubed, $V = l^3$, then $l = \sqrt[3]{N_A \text{ cm}^3} = 8.4 \times 10^7 \text{ cm}$. E1-28 The

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