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## Calorimetry

### **Experimental Chemistry ...**

\* For this calculation, the quadratic formula was used. If one makes the simplifying assumption that  $x$  is small relative to  $[\text{HCOOH}]$ , the calculated value of  $[\text{H}_3\text{O}^+]$  is 0.0013 M. Although a weak acid dissociates only slightly in water, the soluble salt of a weak acid (for example, sodium formate) is a

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strong electrolyte and dissociates completely.

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responses in the following statement:  
Density is a physical / chemical property  
of matter and an intensive /extensive  
property of matter.. What devices will  
you use to measure the mass and the  
volume of water in Part A of this lab?

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analyze and classify the reactions into certain groups.

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If any NaOH spills on you, rinse immediately under running water for up to 15 minutes and report the accident to your instructor. Titration Procedure Your

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instructor will demonstrate the correct use of the volumetric pipette and burette at the beginning of the lab session.

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wavelengths in photons discharged from atoms as they move from high energy to low energy, atomic emission.

Hypothesis If the element (independent variable) is placed in the fire (controlled variable), then the

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