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9.1 Reactions and

Equations MAIN Idea

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Chemical reactions are represented by balanced chemical equations. 9.2

Classifying Chemical Reactions MAIN Idea

There are four types of chemical reactions: synthesis, combustion, decomposition, and replacement reactions.

9.3 Reactions in Aqueous Solutions

MAIN Idea Double-replacement reactions occur between substances

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Chapter 9: Chemical Reactions

CHAPTER Date STUDY
GUIDE Class Chemical
Reactions Section 9.1
Reactions and

Equations In your
textbook, read about
evidence of chemical
reactions. For each
statement, write yes if
evidence of a chemieal
reaction is present.

Write no if there is no
evidence of a chemical
reaction. 2. 4. 5. 6. 8. A

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tomato smells rotten.

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County School
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Chapter 9 141 Section

9.1 Reactions and

Equations pages

282-288 Practice

Problems pages

284-287 Write skeleton

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equations for the following word

equations. 1. Hydrogen and bromine gases react to yield hydrogen bromide.

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Section 9.1 Answer

Key T/F: A chemical

reaction is present if a
tomato smells rotten t

T/F: A chemical

reaction is present if a
drinking glass breaks
into

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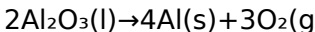
if the percentage yield
for the reaction
represented by the
following equation is

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Chemical

calculated to be 75.3%,
what mass of Al is
expected from the
reaction of 52.5g of
 Al_2O_3



) Aluminum = 26.98

g/mol Oxygen = 16.00

g/mol

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Stoichiometry

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with more related
things as follows types
of chemical reactions

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give you more ideas
and most important:
present you what you
need.

16 Images of Types Chemical Reactions

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Answers**
Worksheets Answers

(a) A straight horizontal line at 1.0; (b) When real gases are at low pressures and high temperatures, they behave close enough to ideal gases that they are approximated as such; however, in some cases, we see that at a high pressure and temperature, the ideal gas approximation breaks down and is significantly different from the pressure

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calculated by the ideal gas equation.

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9 Assessment
Answers**

166 Chapter 9 |
Stoichiometry
Chemical Reactions An
example of a chemical
reaction is the burning

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Chemical

of hydrogen gas (H_2)

to produce steam (hot

H_2O gas). In

chemistry, to burn

something is to react it

with oxygen gas (O_2)

to form one or more

new substances.

Chemical equations are

the language used to

describe chemical

reactions.

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Chemical Reactions

Answers

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Aqueous Solutions •

MHR | 2 Net ionic

equation: $3\text{Ba}^{2+}(\text{aq})$

$+ 2\text{PO}_4^{3-}(\text{aq}) \rightarrow \text{Ba}_3$

$(\text{PO}_4)_2(\text{s})$ Check

Your Solution . The net

ionic equation is

balanced, including the

charges on the ions. 2.

Practice Problem (page

410)

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general chemical
equation for each of
the follow...

**Solved: What is the
general chemical
equation for each of**

...

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The fuel for the space shuttle is hydrogen, which burns in oxygen to produce water vapor and energy. In this chemical reaction, hydrogen is a(n) (9),

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oxygen is a(n) (10),
and water vapor is a(n)
(11). In a chemical
equation for this
reaction, a(n) (12) is
used to separate
hydrogen and oxygen
from water vapor and
energy.

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to the questions and click 'Next' to see the next set of questions. You can skip ...

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...

Engage Discuss the temperature changes in chemical reactions students have conducted so far Chapter 9 chemical reactions answer key. Remind students that

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the decomposition reaction of hydrogen peroxide and the reaction with copper II sulfate and aluminum both caused the temperature of the solution to increase.
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Summary of the Reactions of Chapter 9:

1) Reactions of

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Carboxylic Acids a)

Reaction of a

Carboxylic acid and water. • When placed in water, a carboxylic acid molecule acts as an acid and water acts as a base.

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