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Bohr model of hydrogen Figure 31:
Democritus The atomic theory of matter has a long history, in some ways all the

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way back to the ancient Greeks (Democritus - ca 400 BCE - suggested that all things are composed of indivisible "atoms") From what we can observe, atoms have certain properties and behaviors, which can be summarized as

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Gizmo Answers

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Gizmo Warm-up In 1913, Niels Bohr proposed that the unique spectral lines created by different elements were related to the way electrons were arranged around the nucleus. The Bohr Model: Introduction Gizmo™ explores this connection. The laser shown in the Gizmo can emit photons, or particles of

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light, at a variety of wavelengths.

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Chapter 3 Bohr model of hydrogen
Figure 3.1: Democritus The atomic theory of matter has a long history, in

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some ways all the way back to the ancient Greeks (Democritus - ca. 400 BCE - suggested that all things are composed of indivisible "atoms").

Bohr model of hydrogen - University of Oxford

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absorption spectrum of hydrogen. An

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emission spectrum is created when hydrogen gas emits light. The Bohr model of the atom was inspired by the spectrum produced by hydrogen gas. The link between light and atomic structure is illustrated by the Bohr Model of Hydrogen Gizmo.

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Although revolutionary at the time, the Bohr model is a relatively primitive model of the hydrogen atom compared to the valence shell atom. As an initial hypothesis, it was derived as a first-order approximation to describe the hydrogen atom.

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The Bohr Model | Introduction to Chemistry

Gizmo Warm-up In 1913, Niels Bohr proposed that the unique spectral lines created by different elements were related to the way electrons were arranged around the nucleus. The Bohr Model: Introduction Gizmo™ explores this connection. The laser shown in the

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