

## Advanced Chemistry With Vernier Lab Answers

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Advanced Chemistry with Vernier 19 - 1 Lab: Buffer Preparation and Testing A buffer is a mixture of a weak acid and its conjugate base, or a weak base and its conjugate acid A buffer's function is to absorb acids (H<sup>+</sup> or H<sub>3</sub>O<sup>+</sup> ions) or bases (OH<sup>-</sup> ions) so that the pH of the system changes very, very little In many systems, buffers are critical

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Advanced Chemistry with Vernier 30 - 5 Lab #10 PRESTUDY Gases: Charles' Law Name \_\_\_\_ Section \_\_\_\_ 1)(2 points ) Convert the following temperatures and express the answers to the proper number of significant figures: a) 2530 °C to K b)23 K to °C 2)

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