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AP® Chemistry 2007 Scoring Guidelines The College Board: Connecting Students to College Success The College Board is a not-for-profit membership association whose mission is to connect students to college success and ... Calculate the value of the standard free energy change, ΔD ...

AP Chemistry 2007 Scoring Guidelines - College Board

(a) Calculate the value of the standard free energy change, ΔG° , for the reaction. (b) Determine the temperature at which the equilibrium constant, K_{eq} , for the reaction is equal to 1.00. (Assume that ΔH° and ΔS° are independent of temperature.) (c) Calculate the standard enthalpy change, ΔH° , that occurs when a 0.256 mol sample of NF_3

2007 AP CHEMISTRY FREE-RESPONSE QUESTIONS

The following comments on the 2007 free-response questions for AP® Chemistry were written by the Chief Reader, Eleanor Siebert

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of Mount St. Mary's College in Los Angeles, California. They give an overview of each free-response question and of how students performed on the question, including typical student errors.

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2007 AP® Chemistry Free-Response Questions The following comments on the 2007 free-response questions for AP® Chemistry were written by the Chief Reader, Eleanor Siebert of Mount St. Mary's College in Los Angeles, California. They give an overview of each free-response question and of how students performed on the question,

2007 Ap Chemistry Free Response Questions Form B Answers ...

(c) Calculate the Gibbs free-energy change, ΔG , for the reaction at 862qC. (d) State whether the entropy change, ΔS , for the reaction at 862qC is positive, negative, or zero. Justify

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2007 AP CHEMISTRY FREE-RESPONSE QUESTIONS (Form B)

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AP® CHEMISTRY 2007 SCORING COMMENTARY (Form B)

Question 1 Sample: 1A Score: 9 This response earned all 9 points: 1 for part (a), 2 for part (b), 2 for part (c), 1 for part (d), 2 for part (e), and 1 for part (f). Sample: 1B Score: 6 The point was not earned in part (a) because the student incorrectly attempts to use the Ideal Gas Law to

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Free-Response Questions Scoring Statistics Question Mean Standard Deviation Number of Points Possible 1 4.48 2.86 9 2 3.93 2.81 10 3 3.7 3.11 10 4 7.63 4.46 15 5 3.21 2.34 9 6 3.6 2.37 9

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AP Chemistry Free Response Answers. Adrian has 30 years of high school and early college chemistry teaching experience, in both the UK and the USA.

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AP Chem - 2007 Free Response Q1

Consider the reaction represented above that occurs at 25°C. All reactants and products are in their standard states. The value of the equilibrium constant, K_{eq} , for the reaction is 4.2×10^{17} at 25°C.

Galvanic (Voltaic) Cells

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AP[®] CHEMISTRY 2007 SCORING COMMENTARY (Form B)

Question 1 Sample: 1A Score: 9 This response earned all 9 points: 1 for part (a), 2 for part (b), 2 for part (c), 1 for part (d), 2 for part (e), and 1 for part (f). Sample: 1B Score: 6 The point was not earned in part (a) because the student ...

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2003 Ap Chemistry Free Response Questions Form B Answers

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